Reactive Transport Modeling Tutorial Session #2

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Session objective

- 1. Present challenge of and possible approach to scaling of reactive transport modeling of microbial processes
- 2. Introduce hands on online modeling tutorial with interactive reactive transport modeling
- 3. Discuss and present pipeline for automated microbial modeling using KBASE and PFLOTRAN

Why use reactive transport models for microbial modeling?

- Microbes exist in space and time
- Modeling of microbial processes needs to consider macroscopic spatial and temporal dynamics (fluxes of temperature, moisture and chemicals)
- Reactive Transport models are designed for this

Steefel, de Paolo and Lichtner 2005

Oxidation/Reduction Zones that may develop in aquifer downstream from organic rich landfill:

- Methanogenesis
- Sulfate reduction
- Dissimilatory iron reduction
- Denitrification
- Aerobic respiration



Scaling challenge:

After the workshop, all attendees would like to model microbial dynamics

- At multiple spatial scales
- At multiple temporal scales
- In multiple dimensions
- At multiple locations



There may be millions of models we collectively would like to build and run!

Big question: How do we scale our modeling capabilities?

Mismatch between data generation and model generation capabilities







Data generation

Model generation

Why is scaling hard ?

- Acquiring understanding of what is involved in modeling is complex
- Assembling data for models (geochemical, microbiological, hydrological, environmental) is time and labor consuming
- Building and running microbial models is complex

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What can we do about it?

- Make it easier to acquire a basic understanding of modeling
- 2. Develop and provide data management automation, discovery and access mechanisms
- Create a microbial modeling pipeline that uses KBASE and PFLOTRAN so that we can easily create and run models

Remainder of presentation

1. Introduce online, hands on tutorial which should allow attendants to understand key aspects of modeling

2. Briefly discuss aspects of data management, discovery and access mechanisms

3. Demonstrate results of a microbial modeling pipeline that uses KBASE and PFLOTRAN

Online modeling course

Development motivated by need for educated collaborators -> collaborators do <u>not</u> need to become modelers, but need to understand modeling basics

Key part: interactive, hands on modeling to provide intuitive insights

Low activation energy:

- no need to install PFLOTRAN
- Intuitive GUI

Development of this course was funded under DOE SBIR award DE-SC0015145 by the United States Department of Energy, Office of Science Advanced Scientific Computing Research to Subsurface Insights (PI Dr. Haiyan Zhou)



To get access

- 1. Go to ______ https://learn.subsurfaceinsights.com
- 2. Hit request access button
- 3. Brings up second window
- Enter email, and click request access (make sure to confirm you are not a robot)

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- 5. You will get a registration link
- We will start sending out registration links after going through course overview (so 15-20 minutes into talk)



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Im not a table

Project Contac

Course

- Meant to provide comprehensive introduction to reactive transport modeling
 - General overview of modeling and visualization concepts
 - Chemistry
 - Flow and transport
 - Numerical modeling
 - PFLOTRAN basics
 - Details on PFLOTRAN infile model building
- Throughout course: Interactive PFLOTRAN modeling

Send course comments/suggestions/bugreports to learn@subsurfaceinsights.com

	Learn	Course Overview	About this course	What are numerical models and what do they do?	Course Navigation and Interactive Modeling	PFLOTRAN	Acknowledgements	Disclaimer and Legal Notices	About Subsurface						
	6	Introduction	1												
	Ŋ	Fundamentals of Modeling	Course Na	Course Navigation and Interactive Modeling Depending on the pre-existing knowledge of the student, this course should take between 5 and 15 hours. The material is presented sequentially with modules building on too of one another, but											
		Introduction	students are free t	pre-existing knowledge of the student, this cours o move ahead or circle back. Some details on nav	e should take between 5 and 15 hours. In gation are given below.	e material is pri	esented sequentially with	h modules building on top o	f one another, but						
		Models in Time and Space	odels in Time and Space												
		Example Applications Course navigation Modeling Overview The navigation menu on the left hand side of the course page includes Chapters and Sections. In any given Section, Subsections are in the top menu. Equations and Reactions are numbered throughout the course to aid in finding and referencing these. Equation numbers are in the form "####" and Reactions are designated with "R####" where the numbers are the chapter vection, subsection and exercising subsection and the course to aid in finding and referencing these. Equation numbers are in the form "####" and Reactions are designated with "R####" where the numbers are the chapter vection, and the course to aid in finding and referencing these. Equation numbers are in the form "####" and Reactions are designated with "R####" where the numbers are the chapter vection, and the course to aid in finding and referencing these. Equation numbers are in the form "####" and Reactions are designated with "R####" where the numbers are the chapter vection, and the course to aid in finding and referencing these. Equation numbers are in the form "####" and Reactions are designated with "R####" where the numbers are the chapter vection, and the course to aid in finding and referencing these. Equation numbers are in the form "####" and Reactions are designated with "R####" and Reactions are designated with "R###" and Reactions are designated with "R###" and Reactions are designated with "R####" and Reactions are designated with "R###" and Reactions are designated with "R###													
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		Flow & Transport	The course was te	he course was tested on the latest versions (June 2020) of Chrome and Firefox. It may work on other browsers.											
		Darcy's Law & Fluid Flow	external links												
		Conservative Solute Transport	Hyperlinks to diffe	Hyperlinks to different external resources are provided at multiple places in the text. Two kinds of links are provided: links to other webpages (which should open in a separate window) and links to articles and reports in PDF form. Dependent on your browser configuration clicking the latter link can either open a PDF in a separate window or start download. Where we provide a link to a PDF article and report we indicate this in the text.											
		Coupled Chemistry & Transport	and report we indi												
		Numerical Modeling	Units and I	Dimensions											
		PFLOTRAN	In this course, dim dimensions, the fo	ensions (non-numerical physical properties) will b llowing abbreviations are used:	e specified in square brackets and units (r	ames for meas	urements used on those	dimensions) will be specified	d in parentheses. For						
Santian		Appendix	Mass (M)	Mass (M)											
Section		Options	 Length (L) Time (t) Temperature 	m											
			Glossary of	Symbols											
			Because of the ran	ge of topics covered in this course, which are use	d by a wide variety of fields that often use	different symb	ology, we have included	a Glossary of Symbols in the	e course Appendix. You						

Subsection

Navigation through side menu (sections) and top menu (subsections) should be intuitive

Key piece of course: PFLOTRAN

- PFLOTRAN: open source, massively parallel subsurface flow and reactive transport modeling software
- Developed by DOE scientists at different national labs since 2000
- Simulates transport and reactions in either 0, 1, 2 or 3 D
- <u>https://www.pflotran.org/</u>











PFLOTRAN capabilities

- Transport
 - Saturated and unsaturated flow
 - Multicomponent
 - Mobile/immobile primary species
 - Advection (upwinding)
 - Hydrodynamic dispersion
- Reaction
 - Aqueous speciation
 - Ion activity models
 - General (A + B \leftrightarrow C)
 - Nth order kinetics
 - Reversible
 - Mineral precipitation-dissolution

- Microbiological
 - Michaelis–Menten kinetics
 - Biomass
 - Inhibition
- Radioactive decay with daughter products
- Sorption
 - Isotherm-based: linear, Langmuir, Freundlich
 - Ion exchange
 - Surface complexation
 - Equilibrium
 - Kinetic / multirate kinetic
- Reaction Sandbox

PFLOTRAN is well suited for modeling microbiological processes - challenge is steep learning curve

Solution - expose models through a browser (mechanism used in course)



Architecture allows

- a) Expose model parameters
- a) programmatic model building and modification

Course housekeeping

Once you get an account, for now please turn off Auto-run and sweep models

Send comments/issues to learn@subsurfaceinsights.com

Interface uses currently stable PFLOTRAN version (v2). Will upgrade to v3 in coming month.

Registered users will be on a low volume mailing list with course update information

Appendix	
Options	
□ Auto-run models	
□ Sweep models	

Models and model output basics

Models have dimensionality in space and time:

- we can model in 0,1,2,3 D
- We can model either in time T (1T) or instantaneously (0T)

Models range from 0D, 0T to 3D, 1T

Different models provide different understanding

Key point - understanding model output

- https://learn.subsurfaceinsights.com/learn/#1.1.2.visualization-in-time-and-space
- We can output and visualize in different ways
- Example
 - 1D, 1T model
 - Column with ten blocks of 1 cm each
 - We model the evolution (e.g. thermal diffusion)
- We can look at
 - One location for all times (tec file)
 - All locations for one time (h5 file)
- PFLOTRAN can generate both files



Same model

Different visualization

- Space (top)
- Time (bottom)

Both provide insights



Simplest model: 0d,0-t - Aqueous Complexation

https://learn.subsurfaceinsights.com/learn/#2.1.0.water-chemistry

← → C 🔒 learn.subsurfaceinsights.com/	Jean 192, 1. Davies-chamilety	Detailed PFLOTRAN description
Learn Subsurface Insights	Charge Colt & Bro	befof Versteeg
Learn	CO3++ Concentration: 0.001 (m) CO3++ Concentration: 0.001 (m)	
	If you have parameter sweeps enabled, the model below will show pH results for a range of CO ₂ molalities; if sweeps are disabled it will only show results from the default starting value.	
	Detailed explanation of the PFLOTRAN infile for this model is available in the Appendix (Opens in a new window)	
	Interface Usage Instructions (Automatically opens in a new tab or window)	
	Model in served Ker	
	Carbonation, 00-obs-0.tac 🗸	
	Parameter	
	12 pH obs (1) (0.5 0.5 0.5)*	
	Download file Add to graph 3-3	
	Graph all parameters Clear graph	
	Changer a ans to log scale.	
	Change gi anti to kog inale Mareual stade-	
	Toggle kigend. 4.5	
	-1 -0.3 0 0.5 1	
	"Time $[y]^*$	
	CO3, moleity	
	6.0001	
	• 1e-4 0.1	
	Model name: Carbonation 0D Bun mostel Hede menu	

Instantaneous equilibrium

Aqueous Reactions:

 $2\mathrm{H}^{+} + \mathrm{CO}_{3}^{2-} \rightleftharpoons \mathrm{H}_{2}\mathrm{CO}_{3}^{0} \ (R.\ 3.2.1.6)$ $\mathrm{H}^{+} + \mathrm{CO}_{3}^{2-} \leftrightharpoons \mathrm{HCO}_{3}^{-} \ (R.\ 3.2.1.7)$ $\mathrm{H}_{2}\mathrm{O} \leftrightharpoons \mathrm{H}^{+} + \mathrm{OH}^{-} \ (R.\ 3.2.1.8)$

Primary Species (our choice): H⁺, CO₃²⁻, H₂O

Secondary Species (our choice):

OH⁻, HCO₃⁻, H₂CO₃⁰

Model calculates value of different outputs

Only one value (model is 0d, 0-t)

Not very exciting -> but insightful

Parameter:

"2-pH obs (1) (0.5 0.5 0.5)"	•
"2-pH obs (1) (0.5 0.5 0.5)"	
"3-Total H+ [m] obs (1) (0.5 0.5 0.5)"	
"4-Total CO3 [m] obs (1) (0.5 0.5 0.5)"	
"5-Total H2O [m] obs (1) (0.5 0.5 0.5)"	
"6-Free H+ [m] obs (1) (0.5 0.5 0.5)"	
"7-Free CO3 [m] obs (1) (0.5 0.5 0.5)"	
"8-Free H2O [m] obs (1) (0.5 0.5 0.5)"	

pH for CO3-- molality of 0.026

Interface Usage Instructions (Automatically opens in a new tab or window)



We can also use range of molalities (model sweep)

Increase molality - decrease pH

Is what is expected



Example 2: movement of conservative tracer

https://learn.subsurfaceinsights.com/learn/#3.1.1.advection-&-dispersion

Introduction

This model simulates movement of a conservative tracer $A_{(aq)}$ in a horizontal column 0.04 m in length discretized into 100 grids with unit cross area. The initial concentration of tracer in the column is 0.1 m throughout. There are constant concentration boundaries at either end, set to 1 × 10⁻⁸ m at the left boundary (x=0.0 m) and 0.01 m at the right boundary (x = 0.04 m), as illustrated in Figure 4.2.2.1. The column 'fill' has the properties given in the table below:

Parameter	Value				
Porosity*	1.0 [-]				
Tortuosity	1.0 [-]				
Rock Density	2800.0 [kg/m ³]				
Specific Heat	1000.0 [J/(kg·K)]				
Thermal Conductivity (wet)	0.5 [W/(K·m)]				
Permeability	1.0 × 10 ⁻¹⁵ [m ²]				

* Note that a porosity of 1.0 indicates the column is filled with fluid.





Model makes sense: as we increase the flux velocity, we will decrease the concentration quicker from the initial value of 0.1 M

Note that this is at one location in model, we look over time



Look at model at time 0: concentration is 0.1 M in all models (initial condition)

Interface Usage Instructions (Automatically opens in a new tab or window)



Look at model at time 20K : concentration changes as expected for different flux velocities

Course includes models of increasing complexity

- <u>https://learn.subsurfaceinsights.com/learn/#3.2.0.advection-dispersion-reaction-equation</u>
- <u>https://learn.subsurfaceinsights.com/learn/#5.1.3.example-3---acid-mine-drainage</u>
- + many more
- Allows users to experiment with parameters and get hands on understanding of modeling
- Provides incremental introduction to PFLOTRAN capabilities
- Note 1: Users can download PFLOTRAN files from interface for local modifications or analysis (note: PFLOTRAN v2)
- Note 2: dependent on demand model queue may be big we will monitor this and add resources if needed

Data management, discovery and access mechanisms

- Reactive transport modeling needs more than just microbiological data
- Need easily accessible contextual data for microbial model building
 - Soil temperature
 - Elevation
 - Soil moisture
 - Solar radiation
 - Vegetation cover
 - Riverstage and flow
 - Remote sensing data
 - Snow thickness

. . .





Connecticut River (2 km x 2 km)



- Many data can be discovered through different APIs
- Contextual data use needs data normalization and integration in robust and scalable cyberinfrastructure large part of Subsurface Insights focus

Cloud based PFLOTRAN and web based KBASE can we glue them together to create a microbial modeling pipeline?



Cyber Infrastructure



Cloud based PFLOTRAN and web based KBASE can we glue them together to create a microbial modeling pipeline?



Cyber Infrastructure

PFLOTRAN

Turns out we can!

Nitrogen cycling – a case study



Hanford site river sediment microbe-defined model vs.

Literature-based model

System Definition - Biology

Literature-based Model

- Generic 'Bugs'
 - O Ammonium oxidizing bacteria
 - Nitrite oxidizing bacteria
 - Nitrate reducing bacteria



System Definition - Chemistry

Literature-based Model

• Very limited chemistry

Process	Reactants	Products			
Ammonium Oxidation	NH ₄ ⁺ , O ₂ , CO ₂ , HCO ₃ ⁻	H ⁺ , NO ₂ ⁻			
Nitrite Oxidation	NO ⁻ ₂ , NH ⁺ ₄ , O ₂ , CO ⁻ ₂ , HCO ⁻ ₃	NO ₃ -			
Denitrification	NO ₃ ⁻ , Acetate, H ⁺	HCO_3^-, CO_2^-, N_2^-			

(Liu & Wang, 2012)







Metabolic Modeling in KBase

System Metabolic Reactive Transport Simulation

Raw Output vs Processed

1	id name	formula charge	uptake	take min_uptake		lowerbound		max_uptake		upperbound		
	cpd11576_e0	L-methionine R-	oxide_e0 C5H1	1N035	0			0	-1000	0		
	cpd00001_e0	H20_e0 H20	0	-5.7145	8	-13.095		-100	-3.2678	32	100	
	cpd00023_e0	L-Glutamate_e0	C5H8N04 -1		-0.4006	42	-0.400	542	-1000	-0.040	30642	0
	cpd00971_e0	Na+_e0 Na		0	0	-1000	0	0				
	cpd00058_e0	Cu2+_e0 Cu		0.00050	9282	5.09282	e-05	-100	0.0005	9282	100	
	cpd00067_e0	H+_e0 H		9.15836	2.6499	-100	12	12				
	cpd00107_e0	L-Leucine_e0	C6H13N02	0		0	-2.236	82	-1000			
	cpd00013_e0	NH3_e0 H4N		0	0	0		0				
	cpd00060_e0	L-Methionine_e0	C5H11N02S	0		0		-1000	0	0		
	cpd00322_e0	L-Isoleucine_e0	C6H13N02	0			-2.236	82	-1000	0		
	cpd00637_e0	D-Methionine_e0	C5H11N02S	0		0	0	-1000	0	0		
	cpd00035_e0	L-Alanine_e0	C3H7NO2 0		0	-3.8189	1	-1000	0	0		
	cpd01092_e0	Allantoin_e0	C4H5N403			0		-1000		0		
	cpd00117_e0	D-Alanine_e0	C3H7N02 0		-1.9363	8	-3.818	91	-1000	0		
	cpd00307_e0	Cytosine_e0	C4H5N30 0		0	0	-1000	0	0			
	cpd00254_e0	Mg_e0 Mg		0.00050	9282	5.09282	e-05	-100	0.0005	99282	100	
	cpd00159_e0	L-Lactate_e0	C3H5O3 -1		0	0	-1000	0				
	cpd10516_e0	fe3_e0 Fe		0.00203	713	0.00020	3713	-100	0.0020	3713	100	
	cpd01914_e0	L-Methionine S-	oxide_e0 C5H1	1NO35	0		0	0	-1000	0		
	cpd00162_e0	Aminoethanol_e0	C2H8NO 1		0	0	-1000	0	0			
	cpd00007_e0	02_e0 02		0	0	0		0				
	cpd00156_e0	L-Valine_e0	C5H11N02	0		0	-2.656	22	-1000	0		
	cpd00011_e0	CO2_e0 CO2		-0.01	-8.7929	4	-100	-0.01	-0.01			
	cpd00099_e0	Cle0 Cl	-1	0.00050	9282	5.09282	e-05	-100	0.0005	9282	100	
	cpd00149_e0	Co2+_e0 Co		0.00050	9282	5.09282	e-05	-100	0.0005	9282	100	
	cpd00528_e0	N2_e0 N2		-3.1263	5	-4.9094	5	-100				
	cpd00030_e0	Mn2+_e0 Mn		0.00050	9282	5.09282	e-05	-100	0.0005	9282	100	
	cpd00209_e0	Nitrate_e0	NO3 -1		10	6.18109	0	10	10			
	cpd00048_e0	Sulfate_e0	045 -2		0.03659	94	0.0036	5994	0	0.0365	5994	10
	cpd00029_e0	Acetate_e0	C2H3O2 -1		15	4.37511	0	15	15			
	cpd00009_e0	Phosphate_e0	H04P -2		0.13354		0.0133	547	-100	0.1335	547	100
	cpd00034_e0	Zn2+_e0 Zn		0.00050	9282	5.09282	e-05	-100	0.0005	9282	100	
	cpd00242_e0	H2CO3_e0	CH03 -1		-15.702	:9	-17.15	96	-100			
	cpd00205_e0	K+_e0 K		0.00050	9282	5.09282	e-05	-100	0.0005	9282	100	
	cpd00063_e0	Ca2+_e0 Ca		0.00050	9282	5.09282	e-05	-100	0.0005	9282	100	
	cpd11416_c0	Biomass_c0	null 0		-0.1644	72	-0.164	472	-1000	-0.016	54472	0

NiteStep1 1 biomass compounds omitted from reaction string 10.0 NH4+ + 0.000765313 Mg++ + 0.00306125 Fe+++ + 0.000765313 Co++ + 1.31783 O2(aq) + 0.000765313 Zn++ + 9.73097 HC03- + 0.000765313 Mn++ + 0.054999 S04-- + 0.000765313 Cu++ + 0.200688 HP04-- + 0.000765313 Ca++ + 0.000765313 Cl- + 0.000765313 K+ -> 7.22562 H+ + 7.88192 NO2-YIELD 0.247157 Denite 1 biomass compounds omitted from reaction string 0.000508709 Cu++ + 9.14819 H+ + 0.000508709 Mg++ + 0.00203483 Fe+++ + 0.133397 HP04-- + 0.0365582 S04-- + 0.000508709 Mn++ + 0.000508709 Ca++ + 0.000508709 Co++ + 0.000508709 K+ + 0.000508709 Zn++ + 15.0 Acetate- + 0.000508709 Cl- + 10.0 NO3- -> 0.400191 LGlutamate- + 1.93945 DAlanine(aq) + 3.12583 N2(aq) + 15.7133 HCO3-YIELD 0.164287 NiteStep2 1 biomass compounds omitted from reaction string 10.0 NO2- + 0.00032004 Cu++ + 1.0 NH4+ + 0.00032004 Mg++ + 0.00128016 Fe+++ + 4.06889 HC03- + 0.00032004 Ca++ + 0.00032004 K+ + 0.0839232 HP04-- + 0.0229996 S04-- + 0.00032004 Mn++ + 0.00032004 Co++ + 0.00032004 Zn++ + 0.00032004 Cl- -> 10.1138 NO3-YIELD 0.103357 Ncycle.NS1 Reaction does not proceed



High level agreement, difference in details



Site-specific models account for site-specific chemistry and conditions



Live Example

https://pflotranmodeling.paf.subsurfaceinsights.com/pflotran-simple-model/



Continuous Operation



Sequential Carbon Source Switching



Summary

- Online course should make attendees familiar with modeling basics
- A robust cyberinfrastructure will make your life easier
- KBase PFLOTRAN pipeline allows for comprehensive modeling of subsurface microbiology

Contact Rebecca or Roelof for further details